## Module Handbook

Module Name	Physi	cal Organic Chemis	try			
Module Level	Bachelor					
Kode Mata Kuliah	KIO 301					
Sub Heading, if applicable	-					
Courses included in the	-					
module, if applicable						
Semester/Term	5 / Third Years					
Person in charge	Tjitjik Srie Tjahjandarie, Ph.D/Dr Alfinda Novi Kristanti					
Member	Dr. Mulyadi Tanjung, M.S/Dr Hery Suwito					
Language of instruction	Bahasa Indonesia					
Classification within the	Compulsory Course / Elective course					
curriculum	-	-				
Teaching format / class	3 class hours /week (50 min / hour)					
hours per week during the						
semester						
Workload	every week : 3 hours in class, 3 hours structured activities, 3 hours					
	individual learning, 13 week a semester.					
	Total per semester 117 hours ~ 3.9 ECTS *					
Credit Points	3 SCU					
Requirements	Organic Chemistry II (KIO 203)					
Learning	General Competence (skill) :					
goals/competencies	Students have capability on time management and communication skill					
	to give presentation on specific topic of organic chemistry					
	Specific Competence :					
	After following this Course, students able to evaluate the properties of					
	intramolecular, reactivity based on the structure, reactions and					
	stereochemistry of organic compounds					
Course Content	Intramolecular properties of organic compounds consisting of					
	resonance, Hyperconjugation, covalent bonds, bond energy, bond					
	length, the dipole moment and acid-base equilibrium. The relationship					
	of structure and reactivity of organic molecules based on addition					
	reaction, nucleophilic substitution, electrophilic substitution,					
	elimination and rearrangement reactions. Stereochemistry and					
	reactivity relationshipof molecules based on enantiomer,					
	diastereomer, molecular conformation, absolute configuration,					
	regiospecificand regioselective reaction.					
Study/Exams achievement	Passing grade of the Course is formulated using rubric as follows.					
	Students Grading system:					
	Assigment : 20%					
	Quiz : 20%					
	<ul> <li>IVIIa-semester test : 30%</li> <li>Final compater test : 20%</li> </ul>					
	<ul> <li>Final-semester test : 30%</li> <li>Crading based on university academic rules;</li> </ul>					
	Grading based on university academic rules:					
			Giade			
		75 - 100	A			
		70 - 74,9	AB			
		65 - 69,9	В			

		60 - 64,9	BC					
		55 - 59,9	С					
		40 - 54,9	D					
		0 - 39,9	E					
Learning Methods	Lecturing, discussion, tutorials							
Forms of Media	LCD, computer, whiteboard							
Literature	1.	. Aitken, R.A. and Kilenyi, S.N., 1994, Asymmetric Synthesis, Blakie						
	Academic and Professional, London							
	2.	2. Gould, E.S., 1964, Mechanism and Structure in Organic Chemistry,						
		Holt, Linehart and Winston, New York						
	3.	Issacs, N.S., 1992, Physical Organic Chemistry, 2 <sup>nd</sup> ed., Longman						
	:	Science Technical, John Wiley and Sons Inc., New York						
	4. Sykes, P., 1986, A Guidebook to Mecanism in Organic Chemistry,							
		Longman Group Ltd., Essex						
Notes	*Total ECTS = {(total hours workload x 50 min ) / 60 min } / 25 hours							
	Each ECTS is equals with 25 hours							